

COUNTRY : USSR
 CATEGORY : Cultivated Plants. Legumes, Grasses and Roots.
 ED. JCNM. : IZDANIE, №. 1, 1959, №. 4706
 AUTHOR : Sabachinskij, S.A.
 MFT. : AS Uzbek SSR
 TITL. : Storage of Nitrogen and Phosphorus in Lucerne Roots
 DURING Various Phases of Mowing.
 DRG. PIB. : Uzb. SSR. Fanler Akad. dokusulari, Dushanbe.
 AL'UZSSR, 1947, №. 8, 57-60
 ABSTRACT : At the Central Selection Station of the All-USSR Cotton
 Scientific Research Institute, the greatest accumulation
 of root mass with the highest N and P contents was ob-
 tained under mowing of the lucerne in the mass blooming
 phase (167.5kg/hectare of N and 42.4g/hectare of P_2O_5)
 while the smallest N and P contents was obtained in
 the budding phase (96.3kg/hectare of N and 21.7kg/hectare
 of P_2O_5). The highest lucerne seed crop with a single
 irrigation was obtained under mowing it at the beginning
 of blooming (49.4 centners/hectare of seed or a total

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ZABASHTANSKIY, S.A.

Accumulation of phosphorus and nitrogen in the roots of alfalfa
when it is cut at different stages of growth. Dokl. AN Uz. SSR
no.8:57-60 '57. (MIRA 11:5)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut khlopkovodstva.
Predstavleno chленom-korr. AN UzSSR A.M. Mal'tsevym.
(Alfalfa) (Roots (Botany))

ZABASHTANSKIY, Stanislav Antonovich, kand. sel'khoz. nauk;
NIYAZOVA, R., red.

[Triumphal step of the Bukhara youth; practices of
Nasreddin Pulatov's Brigade on the "Uzbekistan" Col-
lective Farm in Vabkent District, Bukhara Province]
Pobednaia postup' bukharskoi molodezhi; opyt Nasred-
dina Pulatova iz kolkhoza "Uzbekistan" Vabkentskogo
raiona Bukharskoi oblasti. Tashkent, Gos. izd-vo
UzSSR, 1963. 34 p. (MIRA 17:9)

AKHUNOVA, Tursunoy, Tseroy Sotsialisticheskogo Truda; ZABASHEVSKIY,
Stanislav Antonovich; MARTYNOV, Aleksey Nikiforovich;
STEPANOV, M.A., nauchn. red.; TOCHILINA, L.V., red.

[Technology of cotton growing and harvesting] Tekhnologiya
vozdelyvaniia i uborki khlopchatnika. Moskva, Vysshais.
shkola, 1964. 117 p. (MIRA 17:9)

1. Kolkhoz imeni Kirova Yangiyul'skogo proizvodstvennogo
upravleniya (for Akhunova)

RUMANIA

ZAHAVAI, I., Eng., Candidate in Sciences (Candidat in Stiinte), of the "N. Balcescu" Agronomic Institute (Institutul Agronomic "N. Balcescu"), Bucharest.

" 'Batat', a Valuable Fodder Plant."

Bucharest, Revista de Zootehnie si Medecina Veterinara, Vol 13, No 7, Jul 63, pp 22-26.

Abstract: "Batat", or sweet potato -- Ipomoea batatas-Convolvulus batatas -- is a fodder plant of high productivity newly introduced in Rumania since 1954 from the Peoples' Republic of China. The varieties introduced were the "Victoria 100" and the "166". Their botanical characteristics, productivity and nutritional value are described, and the chemical composition of the roots is given. Includes 5 tables.

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RUMANIA

ZABAUA, I., Dr, Eng, of the Agronomic Institute (Institutul Agronomic), Bucharest.

"Some Problems Concerning Feed Digestibility."

Bucharest, Revista de Zootehnice si Medicina Veterinara, Vol 16, No 4, Apr 66, pp 33-37.

Abstract: The author compares digestibility data calculated according to the method of Leroy with the classic values according to O. Kellner. On the basis of calculations for 40 animal fodders, he concludes that the Leroy values show wide deviations from the actual digestibility values and are not suited for practical use in evaluating fodders.

Includes 2 tables.

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ACC NR: AP5025772

SOURCE CODE: UR/0240/65/000/010/0071/0071

AUTHCR: Obaturov, G. M.; Zabavin, A. K.

35

B

ORG: none

TITLE: Determination of summary dose absorption in internal
irradiation (4)

SOURCE: Gigiyena i saniteriya, no. 10, 1965, 71-78

TOPIC TAGS: gamma radiation, radiation drug, ~~radiation effect~~,
isotope, radiation dosimetry, radiation biologic effect

ABSTRACT: In continuation of earlier work, the author presents three formulas for calculating the total dose absorbed by critical organs in internal irradiation, shows examples of the calculations, and compares their advantages and drawbacks. These methods are based on measuring concentrations of radioactive aerosols and the extent of their penetration, the activity of body eliminations, and the radioactivity of the human body. The first formula is based on 2 principles: the change in the amount of radioactive nuclei in the critical organ is equal to their penetration minus their elimination, proportional to N, and, the potency of the dose in the critical organ is proportional to its activity. The

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UDC: 613.648:621.386.82

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ACC NR: AP5025772

formula is further developed for isotopes with various half lives and results are tabulated for the various organs. Its disadvantages are that no account is taken of simultaneous irradiation of other parts of the body, concentrations of aerosols reaching the respiratory tract may vary, and the constants are not sufficiently accurate. The second formula is based on 24 hour elimination for the first day of irradiation. Calculations are correct for short-term irradiation but their accuracy is low for most long life isotopes and the constants have low accuracy. The third method is one in which the dose of internal irradiation is determined in a given case by the results of measuring radioactivity in the organism and by a formula expressing the internal radiation dose. It is accurate except for the coefficients but requires expensive equipment. Orig. art. has: 11 formulas and 4 tables.

SUB CODE: 06 / SUBM DATE: 07Apr64

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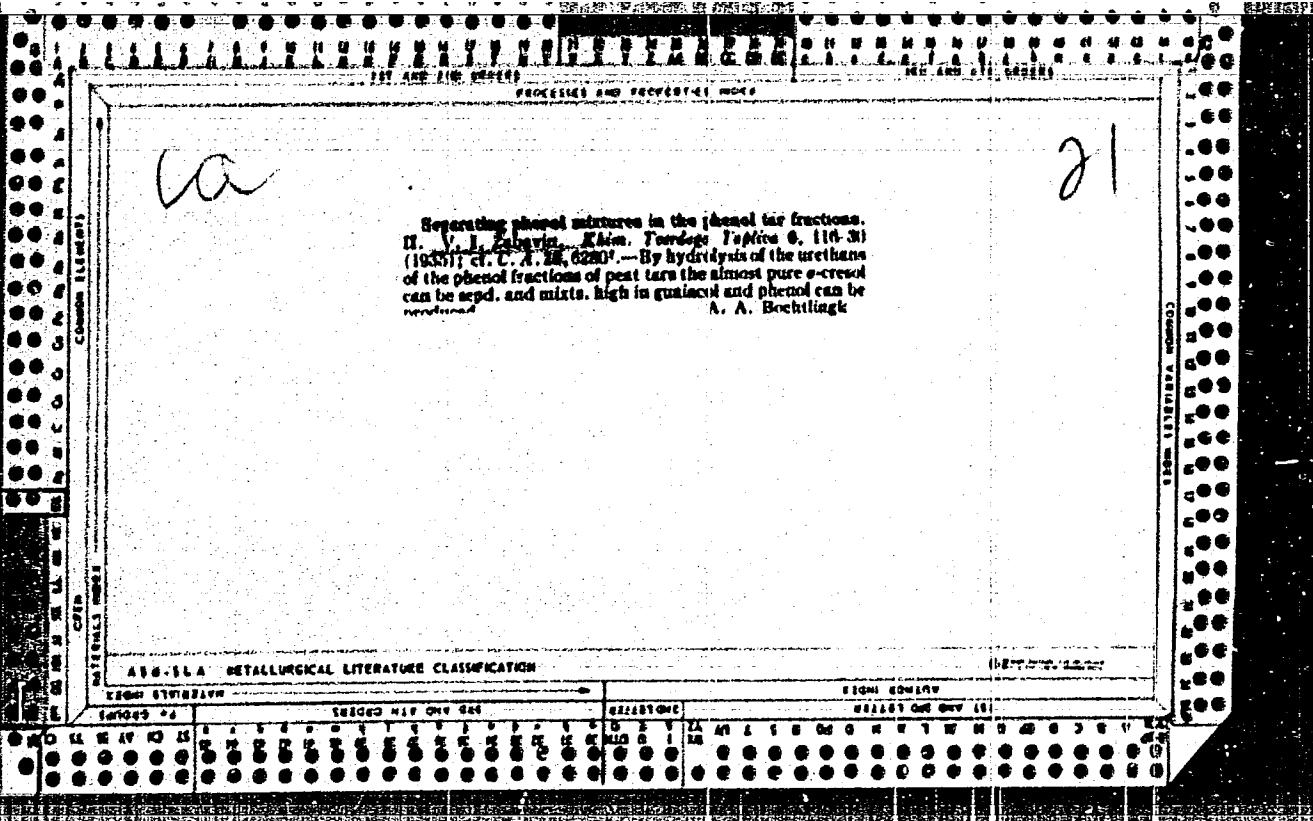
REF. AND ID. NUMBER		PROCESSES AND PROPERTIES INDEX		MATERIALS INDEX	
<p>Conversion of higher molecular homologues. V. I. Kostom (Educa. Tsvet. Vol. 1933, 3, 567-564). Dinitroso hydroquinone (dinitrophenols of low bp) and aromatic hydrocarbons. Yields and type of products depend on duration and temp. of reaction. Cr. Ann.</p>					
ASIN-100 METALLURGICAL LITERATURE CLASSIFICATION					
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Separating mixtures of phenolic and phenol-containing fractions of tar. V. I. Zaitseva, Khim. i Tekhn. Toplits 4, 450-51 (1953).¹-Form carbones were obtained by treating the phenol with Ca(OH)_2 (twice the amt. required by theory) and passing a stream of phenogene until the soln. became clear. A fractional decomps. of phenol carbonate salts was obtained with 25% NH_4OH ; it was found that the segn. of the individual phenols depends on the amt. of NH_4OH and the duration of contact. A. A. Rostovtsev.

A. A. Rocklin

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The substances extracted with petroleum ether from
Prof. V. I. Zalovin, Akad. Nauk SSSR, Teplofizika 9, 108-75
(1960). The total residue obtained from sludge of the
primary peat tar was investigated. Methylation and
ultimate analysis established the humic character of the
asphaltic substances of the tar. The asphaltic substances
ext. from tar should be considered as humic acids of the
initial peat, transformed during dry distillation. Some of the
substances ext. dissolved completely in alkali; their
composition and properties were similar to those of humic
acids. In spite of clearly defined humic characteristics,
all substances ext. from tar dissolved in the org. solvents,
showing the properties of humic acids and bitumen simultaneously.
A. A. Podgorny

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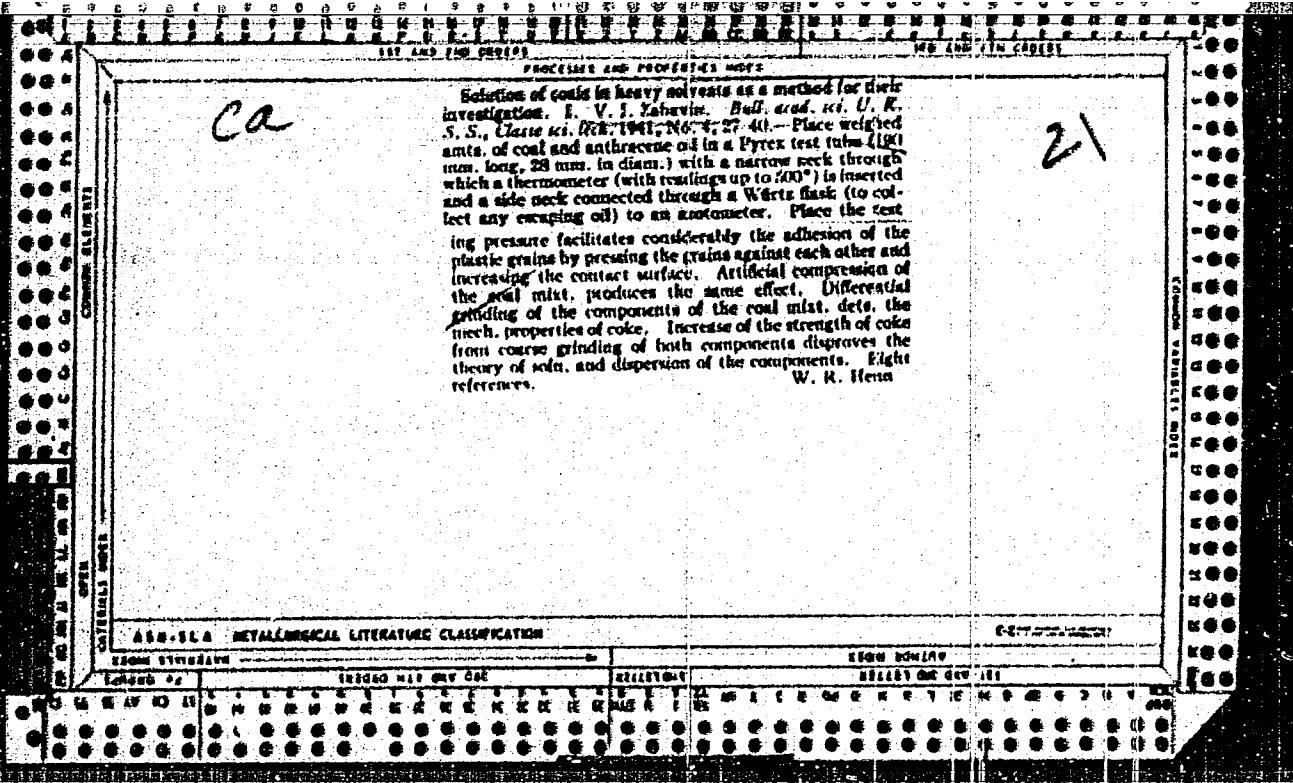
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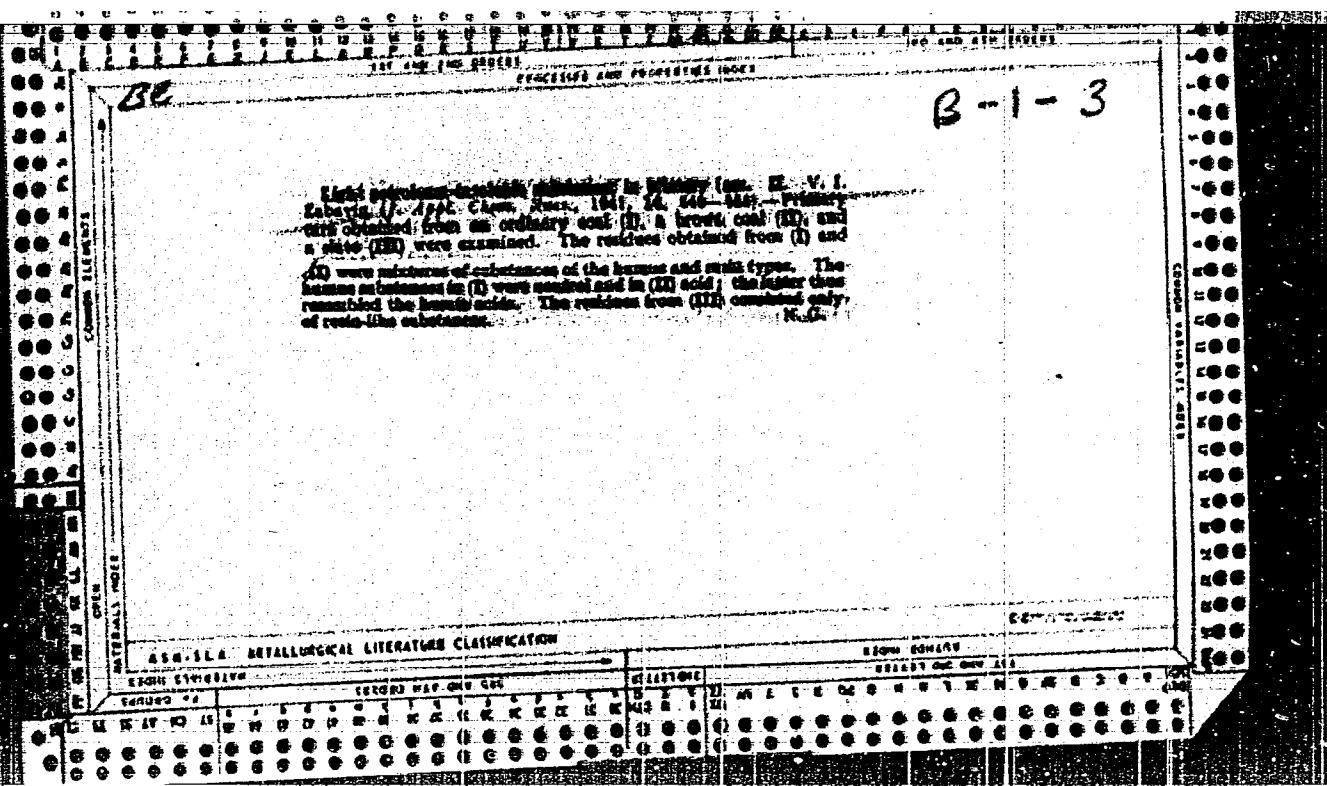


Relation of acids in heavy solvents as a method for their characterization and assay. V. I. Zabotin. *Mull. Acad. U.S.S.R., Classe sci. (tek.),* No. 7, 71-74; *J. C. S. A.*, 16(1927). Ketene (I) dissolves a smaller fraction of coal than anthracene oil (II) (4-40%, instead of 12-81%) but has the advantage of leaving the extract uncontaminated. The solv. in II varies, but in I does not vary, with the coking qualities of coal.

ASA-51A METALLURGICAL LITERATURE CLASSIFICATION

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1858. REACTION OF THE ORGANIC SUBSTANCE OF COAL WITH BENZOIC ANHYDRIDE. Zabarin, V. I. (Bull. Acad. sci. U.R.S.S., C. sci. tech., 1943, Part 8, 35-49; Brit. abstr. 1945, B I 10). When powdered coal is added to boiling Be_2O the b.p. of the liquid gradually falls. The rate of fall and the final b.p. depression are the larger the younger is the coal; they can be used for characterising a coal specimen. The rate of depression rises with the ratio coal: Be_2O ; the ratios 1:20, 1:40, and 1:80 were used. The reaction product contains Be_2O , BeOB , a product sol. in CHCl_3 , and a product insol in CHCl_3 . 1g. of coal yields 0.2-1.7 g. of sol. and 0.74 g. of insol. product. If a coal is first dissolved in anthracene oil, pptd. by C_6H_6 , dried, and then heated with Be_2O , the yield of sol. product is raised and of insol. product lowered. Both these products can be saponified by KOH; the sap. val. is the greater the younger is the coal (0.6-0.3 g. of KOH per 1g. of product). The reaction between coal and boiling Be_2O produces also a little CO_2 , CO , H_2 and paraffinic hydrocarbons, but at 300° a similar reaction occurs without evolution of gas.

ASH-SEA METALLURGICAL LITERATURE CLASSIFICATION

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Oxidation of oils, fats, and asphaltenes of Perm petroleum, and problems of genesis of asphaltic and tarry substances. V. I. Zabavkin. Bull. Acad. sci. U.R.S.S. Classe sci. 1957, No. 9/10, 57-67.—From a sample of crude petroleum an asphaltene fraction (I), a resin fraction (II), and an "oil fraction" (III) still contg. some resins were isolated. Oxidation of I by an air current at 180-80° for 210 hrs. produced a dark, viscous (both chemically and physically) to natural bitum acids. Oxidation II was slower (482 hrs. at 180-8°) in an O current and yielded an asphaltene-like product (IV) which differed from I by its reaction with $\text{MeOH} + \text{HCl}$ and with Cl_2N_2 , and by its resistance to further oxidation. Oxidation of III by O for 250 hrs. at 180-8° gave a product similar to IV, volatile substances, etc. Probably, natural asphaltenes are formed through oxidation and decompn. of other petroleum substances. B. A.

Inst. Mineral Fuels, AS USSR

1957-1960 METALLURGICAL LITERATURE CLASSIFICATION

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Solution in heavy solvents as a method of extraction and investigation of coals. V. I. Salavina and V. I. Barkova. Bull. Acad. sci. U.S.S.R., Chem. sci. 1946, 7(7-8) (in Russian); cf. C.A. 40, 14621; 39, 2400. Coal dissolves to a large extent in aromatic oil (A) (fraction 350-380° from coal-tar) and in xylene (R) (from xylene, m. 12-13°), somewhat below 350°, at ordinary pressure; in this process the constituents of the coal undergo no chemical change. Various Donets-Basin

coals showed solubilities in A of 22-90%; in R of 4-50%; Kuznetsk-Basin coals dissolved in A to the extent of 88-92%, in R to 48-53%; in all cases solv. in A is substantially higher, roughly twice as high as in R. Solv. of coal in pyridine is several times higher after treatment with either A or R; this indicates that solv. in A and R merely changes the colloidal structure of the constituents of the coal (depolymerization) so as to render them more readily sol. in lighter solvents such as pyridine. In this respect, A is more efficient than R; e.g., solv. of a given coal in pyridine was raised from 17% to 64% and 80%, resp., after treatment with R and with A. The coking ability of a coal residue in the fraction sol. in the heavy solvent; the insol. residue does not coke but gives a powder; the latter fraction also contains practically all the ash. The yield in volatile matter from the A-dissolved fraction is equal to, or somewhat lower than, that of untreated samples; the yield from the residue is the same as, or lower than, that from the original coal. With R, the yield in volatile matter is 6-12% higher from the dissolved fraction than from the residue. The lower dissolving power of R as compared with A and the higher plasticity on coking of the R-dissolved fraction indicate that R is more selective solvent toward coal than is A. The R-sol. part is richer in H. This distribution contrasts with

the results of Beausis *et al.*, with toluene at 300° of a coal of similar composition and similar R-value. With the object of doing the nature of the difference in solvent behavior of A and R, samples of coal and its fractions were subjected to oxidation. On exposure to air at 150-180°, for 168 hrs., the R-sol. fraction yield is 1.7 times more H₂O and 1.8 times more C₂H than the insol. residue. The oxidation products contain C 70.34%, H 3.04%, ether-methyl 4.40%, ether-methoxy 0.10%; for the R-sol. fraction and the insol. residue, the figures are 72.32, 3.50, 3.68, 3.92% and 73.25, 2.65, 3.28, 5.21%, resp.; thus all three substances are of the same chem. nature and differ only in colloidal condition. The same results were obtained in oxidation expts. with permanganate. A extract from coal all its plastic constituents; this fraction is further split by R into a hydrohumic est. and a huminic residue. N. Tsch

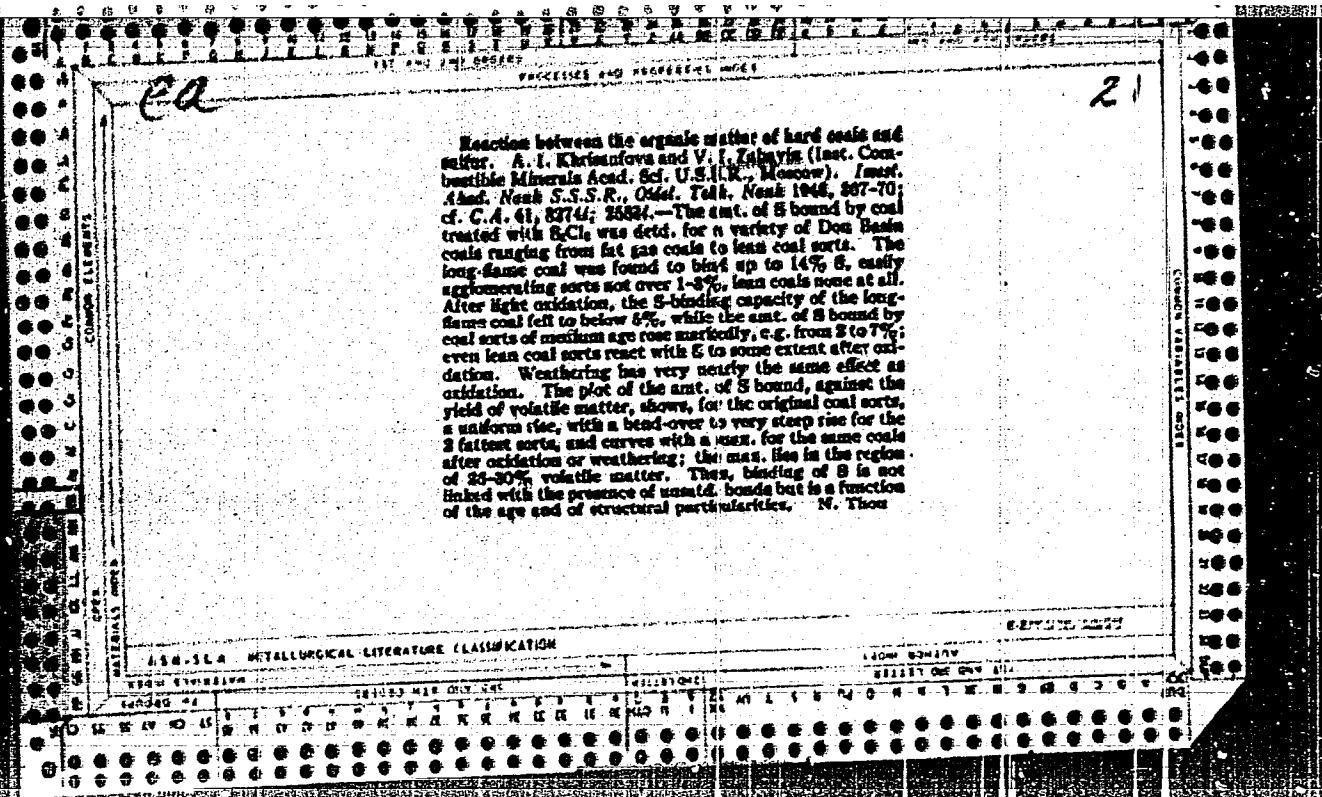
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Reaction between the organic matter of coal and sulfur.
V. I. Sabavin and A. I. Khrisanova. *Bull. Acad. sci. UkrSSR. Ser. Sci. tekhn.* 1946, 1020-34 (in Russian).-

(1) Samples (0.1-0.2 g.) of 11 different hard coals were heated with 3.8 g.owers of S at 300, 400, and 400°. From the calorimetrically dried ash, of 11/1 formed, the amt. of H, split off by the coal was found to increase with temp. in all cases; for example, for 2 different coals 0.63, 1.25, 1.07 and 0.03, 0.13, 0.37%, at 300°, the amt. of H ranged from 0.11 to 1.30%. There is a distinct parallelism between the amts. of H, and the vol% of the coal in anthracene oil and in retene. Reaction of S with coals in these solvents results in increased splitting off of H, nearly the same for most coal varieties: 2.4-2.9% at 350°, and varying much less between 150 and 400°. The retene-sol. fraction splits off, evidently, about 3-3.5% H, the retene-insol. about 2%; the two fractions differ by about 1% H in their elementary组成. Evidently, vol% in retene, resulting in both a disruption of the cellulose structure and of chrm. bonds, reveals original differences between coal varieties. On the other hand, reaction with S permits characterization of the given sort of coal. (2) The amt. of S bound by coal reacting with a benzene soln. of $S\cdot Cl_2$ is mostly 4-8% (in one case 1.2%), with no apparent relation with the original S content of the coal. After vol% in retene and subsequent reaction with $S\cdot Cl_2$, the amt. of S bound by a coal with an original 1.22% S rose from 4.10 to 8.25%, in another instance (originally 2.6% S) from 1.2 to 9.6%. Practically the same amt. of S (8-10%) is bound by the retene-sol. and the insol. fractions. The observed considerable increase of bound S after vol% in retene cannot be reconciled with the representation of Postovskii and Kharlamovich (*C. A.* 41, 2777), according to which the binding to this ether-hydrogen-like addn. of S to double bonds in coal (similar to addn. of S in the vulcanization of rubber), since there is no reason why vol% in retene should increase the no. of double bonds in coal. The S bound by the coal is easily given up, soln. of the sulfurous coal in retene evolving HS until the S content does not exceed that of the original coal.

N. Todor

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<p style="text-align: center;">117 AND 118 CHARTS PRACTICE AND PREDICTIVE MODE</p> <p><u>1660.</u> REACTION WITH ABIEIC ACID AS A METHOD OF CHARACTERIZATION AND INVESTIGATION OF COAL. Zabavina, V. I. (Bull. Acad. Sci. U.S.S.R., Sect. Tech. Sci. July 1967, 871-880). Gives results of investigation of the products of reaction and solution of two types of coals with abieic acid. Use of this acid confirms and extends information concerning coal structures etc. obtained by use of anthracene oil as solvent.</p> <p style="text-align: right;">E. L. R.</p>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																												
<p style="text-align: center;">ASCELS METALLURGICAL-LITERATURE CLASSIFICATION</p> <table border="1"> <tr> <td>GENERAL</td> <td>1</td> <td>2</td> <td>3</td> <td>4</td> <td>5</td> <td>6</td> <td>7</td> <td>8</td> <td>9</td> <td>10</td> <td>11</td> <td>12</td> <td>13</td> <td>14</td> <td>15</td> <td>16</td> <td>17</td> <td>18</td> <td>19</td> <td>20</td> <td>21</td> <td>22</td> <td>23</td> <td>24</td> <td>25</td> <td>26</td> <td>27</td> <td>28</td> <td>29</td> <td>30</td> <td>31</td> <td>32</td> <td>33</td> <td>34</td> <td>35</td> <td>36</td> <td>37</td> <td>38</td> <td>39</td> <td>40</td> <td>41</td> <td>42</td> <td>43</td> <td>44</td> <td>45</td> <td>46</td> <td>47</td> <td>48</td> <td>49</td> <td>50</td> <td>51</td> <td>52</td> <td>53</td> <td>54</td> <td>55</td> <td>56</td> <td>57</td> <td>58</td> <td>59</td> <td>60</td> <td>61</td> <td>62</td> <td>63</td> <td>64</td> <td>65</td> <td>66</td> <td>67</td> <td>68</td> <td>69</td> <td>70</td> <td>71</td> <td>72</td> <td>73</td> 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</table>				GENERAL	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118	119	120	121	122	123	124	125	126	127	128	129	130	131	132	133	134	135	136	137	138	139	140	141	142	143	144	145	146	147	148	149	150	151	152	153	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	169	170	171	172	173	174	175	176	177	178	179	180	181	182	183	184	185	186	187	188	189	190	191	192	193	194	195	196	197	198	199	200	201	202	203	204	205	206	207	208	209	210	211	212	213	214	215	216	217	218	219	220	221	222	223	224	225	226	227	228	229	230	231	232	233	234	235	236	237	238	239	240	241	242	243	244	245	246	247	248	249	250	251	252	253	254	255	256	257	258	259	260	261	262	263	264	265	266	267	268	269	270	271	272	273	274	275	276	277	278	279	280	281	282	283	284	285	286	287	288	289	290	291	292	293	294	295	296	297	298	299	300	301	302	303	304	305	306	307	308	309	310	311	312	313	314	315	316	317	318	319	320	321	322	323	324	325	326	327	328	329	330	331	332	333	334	335	336	337	338	339	340	341	342	343	344	345	346	347	348	349	350	351	352	353	354	355	356	357	358	359	360	361	362	363	364	365	366	367	368	369	370	371	372	373	374	375	376	377	378	379	380	381	382	383	384	385	386	387	388	389	390	391	392	393	394	395	396	397	398	399	400	401	402	403	404	405	406	407	408	409	410	411	412	413	414	415	416	417	418	419	420	421	422	423	424	425	426	427	428	429	430	431	432	433	434	435	436	437	438	439	440	441	442	443	444	445	446	447	448	449	450	451	452	453	454	455	456	457	458	459	460	461	462	463	464	465	466	467	468	469	470	471	472	473	474	475	476	477	478	479	480	481	482	483	484	485	486	487	488	489	490	491	492	493	494	495	496	497	498	499	500	501	502	503	504	505	506	507	508	509	510	511	512	513	514	515	516	517	518	519	520	521	522	523	524	525	526	527	528	529	530	531	532	533	534	535	536	537	538	539	540	541	542	543	544	545	546	547	548	549	550	551	552	553	554	555	556	557	558	559	560	561	562	563	564	565	566	567	568	569	570	571	572	573	574	575	576	577	578	579	580	581	582	583	584	585	586	587	588	589	590	591	592	593	594	595	596	597	598	599	600	601	602	603	604	605	606	607	608	609	610	611	612	613	614	615	616	617	618	619	620	621	622	623	624	625	626	627	628	629	630	631	632	633	634	635	636	637	638	639	640	641	642	643	644	645	646	647	648	649	650	651	652	653	654	655	656	657	658	659	660	661	662	663	664	665	666	667	668	669	670	671	672	673	674	675	676	677	678	679	680	681	682	683	684	685	686	687	688	689	690	691	692	693	694	695	696	697	698	699	700	701	702	703	704	705	706	707	708	709	710	711	712	713	714	715	716	717	718	719	720	721	722	723	724	725	726	727	728	729	730	731	732	733	734	735	736	737	738	739	740	741	742	743	744	745	746	747	748	749	750	751	752	753	754	755	756	757	758	759	760	761	762	763	764	765	766	767	768	769	770	771	772	773	774	775	776	777	778	779	780	781	782	783	784	785	786	787	788	789	790	791	792	793	794	795	796	797	798	799	800	801	802	803	804	805	806	807	808	809	810	811	812	813	814	815	816	817	818	819	820	821	822	823	824	825	826	827	828	829	830	831	832	833	834	835	836	837	838	839	840	841	842	843	844	845	846	847	848	849	850	851	852	853	854	855	856	857	858	859	860	861	862	863	864	865	866	867	868	869	870	871	872	873	874	875	876	877	878	879	880	881	882	883	884	885	886	887	888	889	890	891	892	893	894	895	896	897	898	899	900	901	902	903	904	905	906	907	908	909	910	911	912	913	914	915	916	917	918	919	920	921	922	923	924	925	926	927	928	929	930	931	932	933	934	935	936	937	938	939	940	941	942	943	944	945	946	947	948	949	950	951	952	953	954	955	956	957	958	959	960	961	962	963	964	965	966	967	968	969	970	971	972	973	974	975	976	977	978	979	980	981	982	983	984	985	986	987	988	989	990	991	992	993	994	995	996	997	998	999	1000
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USSR.

X-ray examination of cokes. A. E. Bicler, V. I. Zatsepin, and A. M. Zubko (Central Sci.-Research Inst. Ferrous Met.). Doklady Akad. Nauk. S.S.R., 87, 587-70(1952). Cokes from high-volatile coal (H.09 V.) and low-volatile coal (10.32 V.) were subjected to soaking in a temp. range from 350 to 1200°, and their structural changes were investigated by an X-ray method. Results obtained show that a structural skeleton of coke stuff is not three-dimensional crystals, but that the lattice-like coks are arranged in parallel and arbitrarily oriented to each other. An intensive growth of the lattice resulted due to the carbon which is deposited during the process of the elimination of volatile matter. W. Parafonow

ZAGAVIN, V. I.

Fuels

Dissertation: "Characteristics of the Chemical Nature, Composition, and Structure of Organic Substances of Coal." Dr Tech Sci, Inst of Mineral Fuels, Acad Sci USSR, 30 Mar 1954 (Vechernaya Moskva, Moscow, 17 March 1954)

SO: SUM 213, 20 Sept 1954

ZABAVIN, V.I.

USSR/Chemical Technology - Chemical Products and Their
Application. Treatment of solid mineral fuels

L-12

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 12838

Author : Zabavin V.I., Gordiyenko N.P., Kleymenova L.A.,
Russianova N.D., Surkova V.L., Sharypkina M.Ya.

Title : On Chemical Composition of Coal and Its Change on
Oxidation

Orig Pub : Khimiya, i tekhnol. topliva, 1956, No 5, 23-31

Abstract : Presented are the results of exhaustive "hot" extraction
(in which the sample is heated by solvent vapor) of coal
of different grades from the new deposits of the Kuznetsk
coal fields, unoxidized and of different degree of disin-
tegration, with alcohol-benzene and with 5% solutions of
KOH in alcohol-benzene removes from coal of grade D and
G₂ 3-12% of extract, ~ 1% from coal of grade Zh, and >
0.5% from coal of grades K-TS. Yield of extract from
oxidized coal of grades G₁ and Zh₂ is higher than from

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USSR/Chemical Technology - Chemical Products and Their
Application. Treatment of solid mineral fuels

L-12

Abs Jour : Referat Zhur - Khimiya, No 4, 1957, 12838

the non-oxidized, reaching in the case of strongly oxidized coal 5-6%. Oxidation of coal of other grades does not increase the yields of extract. Yield of aqueous, alkaline alcohol-benzene extract exceeds by several times that of alcohol-benzene extract, while the yield of aqueous alkaline alcohol-pyridine extract is still higher. Content of acid substances in the extracts increases with increase in the degree of oxidation of the coal. It is appropriate to utilize the method of extraction for an evaluation of the extent of oxidation and in the study of the mechanism of coal oxidation.

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ZABAVIN, V.I.

New methods for determining the degree of oxidation and reduction of coals and the quality of coal on the basis of an oxidized sample. Trudy Lab.geol.ugl. no.6:172-182 '56. (MLBA 10:2)

1. Institut goryuchikh iskopayemykh Akademii nauk SSSR.
(Coal--Analysis)

ZABAVIN, V.I.

Multi-conduit stroboscopic automatic recorder. Priborostroenie
no. 9:28-29 S '56. (MLRA 9:10)

(Electronic instruments)

ZABAVIN, V.I.

Multichannel stroboscopic recorders with intermediate carriers. Pri-
berostroenie no.2:14-15 F '57.
(Electronic instruments) (Stroboscopy)

ZABAVIN, V.I.

AUTHORS: Zabavin, V.I. and Kleymanova, L. A. (Moscow). 24-8-9/34

TITLE: Thermohydrolytic splitting of the basic organic mass of hard coal. (Termo-gidroliticheskoye rasschepleniye osnovnoy organicheskoy massy kamennyykh ugley).

PERIODICAL: "Izvestiya Akademii Nauk, Otdeleniye Tekhnicheskikh Nauk"
(Bulletin of the Ac.Sc., Technical Sciences Section),
1957, No.8, pp. 72-77 (U.S.S.R.)

ABSTRACT: In an earlier paper (13) one of the authors expressed the view that the main organic mass of hard coal may have a structure which is characteristic of the high molecular substances of the polymerhomologic type, i.e. it consists of particles of similar chemical composition built up on the same principle and differing from each other solely by the physical properties. It follows that the residue from the alkali extraction must have a composition similar to that of the dissolving coal particles, i.e. it must consist of substances of acidic and neutral character of an equal or similar nature. The here described experiments were based on earlier observations of one of the authors (14,15) that after dissolution of hard coal at 350 C in anthracene oil, retene and colophony oil and separating from the solution by means of benzole or petroleum ether, the solubility of the coal substance in the benzene, pyridine and phenol increases

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Thermohydrolytic splitting of the basic organic mass of hard coal. (Cont.)

to several times the original value. Particularly, the solubility of the grade Δ and Γ coals in pyridine increases from 15-26% to 65-80%. On the basis of the described experimental results a method was developed of thermo-hydrolytic splitting of the main organic mass of hard coals under soft conditions. The method consists in successive heating of the coal and its residues with an α-naphthol at 280 C and with a 5% solution of potassium hydrate in a mixture of alcohol and pyridine at about 90 C and separation from the solution of substances of an acidic and neutral character. By using this method it is possible to bring into solution and to split into chemical components up to 80% of the organic mass of the hard coals Δ, Γ and ΠЖ by treating them four to six times. The main mass of the investigated coal splits as follows: Grade Δ coal decomposes solely in substances of an acidic nature; coal of the Grade Γ-Δ decomposes into substances which are acidic and neutral in equal quantitative ratio; coal of the Grade ΠЖ decomposes into a substance which is purely neutral in its nature. There are 4 tables, 1 figure and 17 references, 9 of which are Slavic.

SUBMITTED: April 8, 1957.

AVAILABLE: Library of Congress

Card 2/2

ZABAVIN, V.I.

AUTHOR: ZABAVIN, V.I. 106-5-11/13
TITLE: Forming Schemes of Impulses from Sinusoidal Voltage with a Downward-Transformed Feed Voltage. (Schemy formirovaniya impul'sov iz sinusoidal'nogo napryazheniya s ponizhennym napryazheniyem pitaniya, Russian)
PERIODICAL: Radiotekhnika, 1957, Vol 12, Nr 5, pp 73-77 (U.S.S.R.)
ABSTRACT: Schemata are investigated which offer a number of advantages compared to those with multivibrators and trigger systems. The schemata mentioned here have been worked in form of three varieties: with electron tubes, as a combination of electron tubes with semiconductor triodes, and with semiconductor triodes. It is shown that in order to form impulses from a sinusoidal voltage by using condenser discharge systems by means of an electron tube and by using a positive back-feed, it is necessary to create certain conditions in order to prevent parasitical excitation. Such conditions are warranted by the application of a reduced feed voltage of the discharge tube anodes and by the selection of a sufficiently great time constant of the condenser charge. (With 5 Illustrations).
ASSOCIATION: Not given
PRESENTED BY:
SUBMITTED: 4.4.1956
AVAILABLE: Library of Congress
Card 1/1

ZABAVIN, V. I.

AUTHORS: Zabavin, V. I., and Nentsova, V. G. 24-1-14/26

TITLE: Determination of the degree of oxidation of hard coal from the yield of water and of CO₂ during heating. (Opredeleniye stepeni okislennosti kamennyykh ugley po vkhodu vody i uglekisloty pri nagrevanii ugley).

PERIODICAL: Izvestiya Akademii Nauk, Otdeleniye Tekhnicheskikh Nauk, 1958, No.1, pp. 107-112 (USSR).

ABSTRACT: A method expressing correctly the degree of oxidation of hard coal must express the change taking place during oxidation in the entire organic mass of the coal. This requirement is met by methods based on determining the content in oxidized hard coal of oxygen containing functional groups; these methods include the new method described in this paper. The method was developed on the basis of the conception of the primary oxidation of the coal, namely, transformation of the fundamental organic mass of the coal into humic acid and as the limit of oxidation of coal its full transformation into such acid was considered. This assumption of the primary oxidation of coal permits comparison with the limit oxidation of lower stages of oxidation and to express the degree of oxidation by means of a relative

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24-1-14/26

Determination of the degree of oxidation of hard coal from the yield of water and of CO_2 during heating.

number. For developing the practical part of the method, the thermal instability of humic acids was applied. G. Stadnikoff et alii (Ref.8) have shown that humic acid separated from Ukrainian brown coal decomposed during heating, emitting water and CO_2 , owing to breaking up of hydroxyl and carboxyl groups; heating of the acids to 300-350°C resulted in almost complete destruction of the carboxyl groups. Therefore, it could be assumed that the humic acids formed during oxidation of hard coal could also be decomposed during the heating of the coal to 300-350°C, accompanied by the formation of water and CO_2 . The remaining coal substance, which did not yet become transformed into humic acid, will decompose in a similar manner due to the fact that functional groups form in it. On the whole, the separation of water and of CO_2 from the coal will be the more pronounced the more intensive the oxidation of the coal. The greatest yield is obtained during full initial transformation of the organic mass into humic acids. If this assumption Card 2/4 is correct, the quantitative determination of the yield

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Determination of the degree of oxidation of hard coal from the yield of water and of CO_2 during heating.

of water and of CO_2 forming on heating of oxidized coal to 300-350°C permits expressing the degree of oxidation of the coal. The experiments described in this paper confirm these assumptions; they yield results which express satisfactorily the degree of oxidation of the coal and enable development of a simple and rapid method of determining the degree of oxidation. By means of the described method, the yield of the water and of CO_2 is measured, from which the yield of these products from the non-oxidized coal is deducted and the difference is related to the respective yield values from humic acids. A method was developed for determining the degree of oxidation of hard coal from the quantity of water and CO_2 produced by the coal on heating to 350°C. As a measure of the degree of oxidation of the coal, the ratio of the produced water and CO_2 to the quantities produced under equal conditions from coal oxidized in humic acid (and considered as being the limit of the primary oxidation of coal) is applied; the degree of oxidation being expressed in percent. This method produced results which express more accurately the degree

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24-1-14/26

Determination of the degree of oxidation of hard coal from the yield of water and of CO_2 during heating.

of oxidation of the coal than other chemical methods. From the experimental point of view, the main feature of the method is its simplicity and ease and speed of execution.

There are 2 figures, 1 table and 9 references - 8 Russian, 1 German.

SUBMITTED: May 15, 1957.

AVAILABLE: Library of Congress.

Card 4/4

119-3-12/14

AUTHOR: Zabavin, V. I.

TITLE: Tensometric Device With Phase Impulse Modulation
(Tenzometricheskaya ustanova s fazoimpul'snoy modulyatsiyey)

PERIODICAL: Priborostroyeniye, 1958, Nr 3, pp. 29-30 (USSR).

ABSTRACT: The newly developed device the basic diagram of connections of which is given is designed as a Wheatstone bridge with a tensometer in one diagonal serving as transmitter. A sinusoidal phase modulated voltage is used as input. This is new as compared to other similar devices since up to now only amplitude modulated voltages have been used. The peculiarity of the phase modulated voltage consists in the fact that it is due to the addition of two voltage vectors shifted against each other by 90° . The amplifier contains the valves RV 12 P 2000 and has an amplification factor of approximately 1000. The smallest voltage amplitude, which can be transmitted by the potentiometer is 3 mV. The amplifier measures $50 \times 50 \times 100$ mm. Since no special constancy is required from the amplifier it may be constructed also with crystal triodes.

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Tensometric Device With Phase Impulse Modulation

119-3-12/11

There is 1 figure.

AVAILABLE: Library of Congress.

1. Tensometric device--Development 2. Phase modulation
 --Applications

Card 2/2

ZABAVIN, V. I. (Moskva); KLEYKHOVA, L.A. (Moskva)

Characteristics of products derived from the thermal-hydrolytic splitting of the basic organic mass of coal. Izv. AN SSSR. Otd. tekhn. nauk Met. i topl. no.1:90-93 Ja-F '59. (MIRA 12:6)
(Hydrolysis) (Coal)

ZABAVIN, V.I.

TABLE I BOOK EXTRACTIVE

217

Amosova, N.M. <u>Soviet combustible hydrocarbons</u> . Gosudarstvennoye izdatelstvo gosizdat (General or Soviet Books) Moscow, 1961. S. G. Filov, Doctor of Chemical Sciences; Ed. of Publishing House A. Z. Babitskaya; Tech. Ed.: I. P. Rakhuba.	300 p. Printed only (unprinted). 2,000 copies printed.	
YUDOVICH, <u>Fiziko-khimicheskiye oznacheniya na p. 21</u> . Bogorodskoye izdatelstvo.		
BURG, M.M. <u>Karyotyza, Corresponding Member, USSR Academy of Sciences, and Gosudarstvennoye izdatelstvo gosizdat (General or Soviet Books) Moscow, 1959. Doktorant; Tech. Ed.: I. P. Rakhuba.</u>		
YUDOVICH, <u>Fiziko-khimicheskiye oznacheniya na p. 21</u> . Bogorodskoye izdatelstvo.		
<u>DISCUSSION</u> : This collection of articles is intended for geologists, geochemists, and other specialists interested in the genesis of solid mineral fuels. <u>DISCUSSION</u> : The collection of papers on the genesis of solid mineral fuels has been prepared for presentation at the 2nd All-Union Conference on this subject. The formation of humic acids and peat from the decomposition of macro-elements and lignite is discussed in connection with studies on the origins of hard coal and brown coal, and on the role of certain mineral components in the coal- forming processes. The chemical composition of peat and the organic matter of coal is analyzed and shown to a number of substances. Aliphatic hydrocar- bon derivatives are analyzed as are the hydrocarbons of the tar products. Heterocyclic and carbonyl compounds found in different types of coal and the uranium field are also discussed. The transformation of organic matter into combustible minerals is analyzed. The transformation of individual articles.		
<u>DISCUSSION</u> : <u>Kh. T. Qasimov et al.</u> <u>Kazakhstan Oil Shale</u> <u>DISCUSSION</u> : <u>Pozdnyakov, A.B.</u> <u>On the Question of the Origin of Baltic Shale</u> <u>DISCUSSION</u> : <u>Kh. T. T. and I. A. Vilman.</u> <u>Lignite and Bituminous Coals</u> <u>DISCUSSION</u> : <u>I. F. Slobodtsev.</u> <u>Origin of Brown Coal Found in the Subpermafrost Zone of the Urals</u> <u>DISCUSSION</u> : <u>Ivanov, I.A.</u> <u>Geological Carbonization of Sedimentary Coal Found on the Western Flank of the Central and Northern Urals</u> <u>DISCUSSION</u> : <u>I. I. Bogolyubova.</u> <u>Physical and Chemical Characteristics of Some Types of Coal From Yekaterinburg and Magnitogorsk Deposits</u> <u>DISCUSSION</u> : <u>N.I.</u> <u>Conditions of Formation of Bituminous Carbonized Coal From Northern Urals Brown Coal Beds</u> <u>DISCUSSION</u> : <u>V. A. Mironov.</u> <u>Properties of Brown Coal From Pegezhinskoye and Tashchinskoye Deposits on the Southern Flank of the Northern Urals</u> <u>DISCUSSION</u> : <u>A. I.</u> <u>Geological Conditions of Transformation of Coal Shales in the Southeastern Part of the Trans-Ural Platform</u> <u>DISCUSSION</u> : <u>M. Yu. Danilov.</u> <u>Some Unstable Conditions Under Which Coal Shales Can Turn Into Solid Carbon at the Kuznetsk Basin</u> <u>DISCUSSION</u> : <u>D. Z.</u> <u>Formation of Hard Coal During Metamorphism</u> <u>DISCUSSION</u> : <u>I. I.</u> <u>Changes in Microscopic Characteristics of Coalish Coal at Various During Metamorphism</u> <u>DISCUSSION</u> : <u>I. I.</u> <u>Genesis of Turbid Coal at Some Centers</u> <u>DISCUSSION</u> : <u>I. I.</u> <u>Organic Matter in Coal</u> <u>DISCUSSION</u> : <u>V. I.</u> <u>Some General Physical and Chemical Properties of Lignite</u> <u>DISCUSSION</u> : <u>V. I.</u> <u>Some General Properties of Lignite</u> <u>DISCUSSION</u> : <u>V. I.</u> <u>Characteristics of the Process of Transformation of Peat Matter Into Present Combustible Material and the Characteristics of those Characteristics With the Principal Properties of Coal</u> <u>DISCUSSION</u> : <u>I. I.</u> <u>Genetic Features of the Coal Substrates as Ascertained by Diatropographic Methods</u> <u>DISCUSSION</u> : <u>I. I.</u> <u>Chemical Features of the Basic Organic Matter of Hard and Metamorphic Non-Chemical During Metamorphism</u> <u>DISCUSSION</u> : <u>T. A.</u> <u>Changes in the Structure and Properties of Hard Acidic Shales and Gneissic Rocks</u> <u>DISCUSSION</u> : <u>I. G.</u> <u>Role of Mineral Elements in the Coal-Carboniferous Tectonics</u> <u>DISCUSSION</u> : <u>T. G., A. I.</u> <u>Rakhuba, and A. Z. Troparev.</u> <u>General Organic Polycyclic Compounds Contained in Coal</u>		

C L 10319-66

ACC NR: AP5021827

SOURCE CODE: UR/0356/65/000/008/0039/0046

AUTHORS: Kirpichnikov, Ye. (Engineer); Zabava, V. (Correspondent of the journal)

ORG: Leningrad Regional Combin "Lenvodstroy" (Leningradskiy oblastnoy trest
"Lenvodstroy")

TITLE: News in land reclamation technology

SOURCE: Tekhnika v sel'skom khozyaystve, no. 8, 1965, 39-46

TOPIC TAGS: land reclamation, construction machinery, drainage system, agriculture/ E 352 excavator, KM 1400 digger, D 20B grader, DN 1.8 turf cutter

ABSTRACT: Advances in land reclamation technology are discussed. Since the formation of "Lenvodstroy" six years ago, the number of land reclamation stations has grown from 7 to 17. Each station has operational divisions with annual budgets of 250--300 000 rubles. Monthly premiums are paid to workers, technicians, and administrators for exceeding quotas of reclaimed land. The stations have added 8260 hectares of reclaimed land to this region in the past year (almost twice the 1962 figure) with 13 200 hectares anticipated for the present year. Land

UDI: 631.6:626.86

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L 10319-66
AOC NR: AP5021827

reclamation consists primarily of building a network of open drying canals or closed drainage channels. Instead of using excavators E-352 (170 m³ daily capacity) for shallow canal construction, a combination of three machines, namely, canal digger KM-1h00, heavy grader D-20B, and a shovel-scraper, has been found more productive (600 m³ daily). Because of the increasing maintenance cost of the drainage network (148 500 rubles in 1961, 179 800 in 1962, and 463 000 in 1962 for this region), more emphasis has been placed on the quality and life of the canals, resulting in between-maintenance periods of 10--12 years instead of 1--2 years for earlier construction techniques. Water erosion has been lessened by using different methods of slowing water flow (waterfalls, barriers, etc) and by lining the canals (with turf, etc). A new turf cutter developed for this purpose is described in some detail (cutter DN-1.8). Closed drainage ditch excavation has been improved by modernizing excavators ETN-171 and ETN-1h2 for automatically controlled operation (ETN-1h2A). Other labor saving methods, such as streamlined loading, unloading, and continuous laying of drainage pipes, semi-automatic pipe drilling (for joints) (a complete description of a drilling rig is presented), have substantially increased output. Some increased output indicators (amounts of vegetables and potatoes) are tabulated for several combines. Orig. art. has: 6 figures and 1 table.

SUB CODE: 13/
M 02
Card 242

SUEM DATE: none

ZABAVIN, Vladimir Ivanovich; KARPOVICH, V.L., red.

[Bituminous and brown coal; chemical composition and
structure, properties, genesis] Kamennye i burye ugli;
khimicheskii sostav i struktura, svoistva, genezis.
Moskva, Nauka, 1964. 197 p. (MIRA 17:8)

ZABAVIN, V.I. (Moskva); KLEYMENOVA, L.A. (Moskva); STREL'TSOVA, N.T.
(Moskva)

Hydrolytic and thermohydrolytic splitting of the Donets Basin
main organic coal mass. Izv. AN. SSSR. Otd. tekhn. nauk Met. i
tepl. no.2:170-172 Mr-Ap '61. (MIRA 14:4)
(Donets Basin--Coal)
(Hydrolysis)

NAZAROVA, N.I.; MAKAYEVA, R.I.; ZABAVIN, V.I.

Tendency toward the self-oxidation, spontaneous heating, and
self-ignition of the coals of Kirghisistan fields. Iss. AN
Kir. SSR. Ser. est. i tekhn. nauk 2 no.5:9-20 '60.
(MIRA 13:9)

(Kirghizstan-coal)

ZABAVIN, V.I. (Moskva); NEFTSOVA, V.G. (Moskva)

Thermal hydrolysis and hydrolysis in splitting the basic organic
lignite mass. Izv. AN SSSR, Otd. tekhn. nauk Met. i topl. no.2:
168-172 Mr-Ap '59.
(Coal geology) (Hydrolysis)

SOV/180-59-1-17/29

AUTHORS: Zabavin, V.I., Kleymenova, L.A. (Moscow)

TITLE: Characteristics of the Products of Thermo-Hydrological Decomposition of the Main Organic Mass of Brown Coal
(Kharakteristika produktov termo-gidrologicheskogo rasschepleniya osnovnoy organicheskoy massy kamennyykh ugley)

PERIODICAL: Izvestiya Akademii nauk SSSR, Otdeleniye tekhnicheskikh nauk, Metallurgiya i toplivo, 1959, Nr 1, pp 90-93 (USSR)

ABSTRACT: Experiments were made, the results of which are entered in Table 1, p 91; these characterise the products of hydrolytic and thermo-hydrolytic decomposition of the organic mass of the coals D, G and PZh of Kuzbass origin. On the basis of the obtained results, the following conclusions are arrived at: 1) Decomposition products of acidic coals of the grades D and G consist mainly of compounds containing about half as much phenol hydroxyl and about one third to one quarter of carboxyl groups, as humic acids of brown coals. A quantitatively smaller part of decomposition products of acidic coals have a content of functional groups which is near to that contained in humic acids. 2) The decomposition products of G and PZh coals which are neutral also contain

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SOV/180 .59-1-17/29

Characteristics of the Products of Thermo-Hydrological Decomposition
of the Main Organic Mass of Brown Coal

functional groups, but in a quantity which is still smaller than the corresponding acidic products. The content in these of phenol hydroxyls is three to eight times lower than in humic acids of brown coal and the content of carboxyl groups is seven to nine times lower.

3) The molecular weight of the products of hydrolytic decomposition of coals determined cryoscopically is low, and fluctuates between 166 and 650. The equivalent weight of the acidic products varies between 199 and 335, and of the neutral products, between 382 and 1724.

4) The above enumerated indices vary regularly during the process of metamorphosis of the coals D, G and PZh; particularly, the molecular weight of the products of hydrolytic decomposition changes in one direction whilst the equivalent weight changes in the opposite direction.

5) The contents of C, H and N in the products of hydrolytic decomposition of coals of the grades D, G and PZh is lower and the oxygen content is higher, than in

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SOV/180-59-1-17/29

Characteristics of the Products of Thermo-Hydrological Decomposition
of the Main Organic Mass of Brown Coal

the initial coals. The oxygen contained in the investigated coals is between 25 and 39% in the form of functional groups, and between 61 and 75% in another form.

Card 3/3 There are 2 tables and 7 references, 4 of which are Soviet, 2 English and 1 German.

SUBMITTED: February 6, 1958

ZAHAVIN, V.I.; HOROZDINA, L.A.; KHMITSOVA, V.G.

Studying the oxidation process of coals as related to their
tendency for self-heating and self-ignition. Trudy IGI 8:198-212
(MIR) 13:1
'59.

(Coal weathering)

L 31769-66 T/ENP(t)/ETI IJP(c) JD

ACC NR: AP6021700

SOURCE CODE: CZ/0032/66/016/001/0041/0044

23
E

AUTHOR: Zabavnik, B. (Engineer)

ORG: Institute of Technology, Kosice (Vysoka skola technika)

TITLE: Effect of the composition of the mixed nitriding atmosphere upon the thickness and hardness of the hardened surface layer

10

SOURCE: Strojirenstvi, v. 16, no. 1, 1966, 41-44

TOPIC TAGS: nitridation, ammonia, hardness

ABSTRACT: The results are presented of experimental research work on the effect of the proportion of NH₃ in mixed nitriding atmospheres of the NH₃ + N₂ type upon the structure and hardness of the surface layer, comparing two different brands of steel. The optimal composition of the combined atmosphere is given from the viewpoint of the high quality of the surface layer. The recommended compositions are also very economical in terms of production cost. Orig. art. has: 2 figures and 3 tables.

[Based on author's Eng. abst.] [JPRS]

SUB CODE: 11, 07, 20 / SUBM DATE: none / ORIG REF: 002 / SOV REF: 003
OTH REF: 001

UDC: 621.785.53

Card 1/1 (B)

AUTHOR:

Zobovlev, Yu. I.

TITLE:

On the compression of rectangular plates with initial bending

PUBLISHER: Sverdlovskiy zhurnal, Mekhanika, no. 1, 1963, 12, abstract 791
Sverdlovskiy gosudarstvennyi in-t po fiziko-mekhanicheskym issledovaniyam, no. 9, 1961, 43-51

The author has conducted a series of tests of the deformation of a resilient plate with a rectangular cross-section under compression. The plate is made of a thin sheet of rubber. The outer edges of the plate are held by four rectangular specimens which are fastened to the plate. The outer edges of the plate are held by four rectangular specimens which are fastened to the plate. A variant method is used in the solution of the problem. The plate is divided into a number of equal squares. The boundary conditions are determined by the values of the deflection at the vertices of the squares. The problem is solved with known solutions. L. V. Zobovlev

Approved for Release on 03/15/2001 CIA-RDP86-00513R001963310018-9

Card 2/1

- 161-3-63

sup(+) /sup(+) /sup(+) /sup(3)

AESTHETIC

TG
352.153/500 22.10.21

AUTHORS

Zapavnikov, B. S.

TITLE:

The "clap" of rectangular plates with initial distortion

TITLE: The ~~use~~
PERIODICAL: Referativnyy zhurnal Mekhanika, no. 4, 1963, 14, abstract 4V108
(Nauchn. tr. Voronezhsk. inzh.-stroit., in-t, no. 9, 1962, 303-323)

(Nauchn. tr. Voronezhsk. Inst.)

Method of calculating plates having initial bending
and compression of the central part, and with two free sides.
The theory of plates of the third order, and with the following
boundary conditions: free edges, and with the following
boundary conditions: free edges.

For an object to move, it must remain linear, the tangential forces

... The author is also approaching the UIC as a potential patron for the meeting.

surface. He obtains a final equation which depends on the width of the load of q . Czerny also gives

surface. He obtains a final equation involving the load of q^* , Czerny's constant, supplemental sags, zeta and ζ_0 , and of the load of q^* . [Abstracter's note: Complete translation.]

supplemental says, etc.
[Abstracter's note: Complete translation.]
Card 1/2

Card 17-2

APPROVED FOR RELEASE: 03/15/2001

CIA-RDP86-00513R001963310018-9"

16748-63

8/124/63/000/100/042/064

The "clap" of

(zeta plus zeta sub o) (zeta plus 2 zeta sub o) equals q*. He determines values for the coefficients ζ , B for various cases of support, and investigates the condition $\zeta = \zeta_0$. He compares his solution to this problem with earlier ones. A. S. Vol'mir.

Card 2/2

ZABAVNIKOV, B.I., inzh.

Bending of rectangular flexible plates. Trudy MIIT no.122:
328-342 '59. (MIRA 13:5)
(Elastic plates and shells)

ZABAVNIKOV, B.I., insh.

Using the power method in the theory of elastic plates. Trudy MIR
108:308-326 '59
(Elastic plates and shells)

ZABAVNIKOV, B.I., inzh. (Voronezh)

Stability "in the large" of rectangular elastic plates with
initial curvature. Issl. po teor. sooruzh. no.13:189-202
'64. (MIRA 13:2)

ZABAVNIKOV, N.A., kandidat tekhnicheskikh nauk.

Geometric efficiency coefficient and computation of the geometry of toroid
infinitely variable transmission. Avt.trakt.prom. no.9:17-20 S-53.
(MERA 6:9)

1. Moskovskoye vyscheye tekhnicheskoye uchilishche im. Barnana.
(Automobiles--Transmission devices)

ZABAVNIKOV, N.A., kandidat tekhnicheskikh nauk.

On the rational geometry of toroid stepless transmissions. Avt. trakt. prom.
no.11:9-11 N '53. (MLN 6:11)

1. MVTU im. Baumana.

(Automobiles--Transmission devices)

ZABAVNIKOV, N.A., kand.tekhn.nauk

Analytical determination of the starting time and distance.
Avt.prom. 27 no.6:11-14 Je '61. (MIRA 14:6)

1. Moskovskoye vyssheye tekhnicheskoye uchilishche imeni Baumana.
(Automobile engineering)
(Automobile Dynamics)

ZABAVNIKOV, N.S., uchastnik Oktyabr'skoy revolyutsii, personal'nyy pensioner,
chlen Kommunisticheskoy partii Sovetskogo Soyuza s 1917 g.

Role of the telecommunication workers in the Great October
Revolution. Vest. sviazi 22 no.11:26-28 N '62. (MIRA 16:12)

NADEZHINA, Ye.D.; YUDINA, V.V.; ZABAVNIKOVA, N.I.

Accessory sphene from metasomatic trap rocks in the Siberian
Platform (Bol'shaya Botuobiya Valley). Trudy Min. niz. no.14:
243-249 '63. (KIRA 16:10)

(Ulakhan-Botuobuya Valley--Sphene)
(Ulakhan-Botuobuya Valley--Rocks, Igneous)

ZABAVNIKOVA, N.I. Cand Geo Min Sci -- (diss) "Isomorphic
replacements in sphenes of various ^{COMBINATIONS} ~~combinations~~ of the
Soviet Union, Mos 1958, 16 pp with graphs (Acad Sci USSR
^{of Ore Deposits,}
Inst of Geology, Mineral Beds, Petrography, Mineralogy,
and Geochemistry) 125 copies (KL, 21-58,88-89)

- 14 -

ZAHAVNIKOVA, N.I.

Isomorphic replacements in sphenes [with summary in English].
Geokhimia no.3:226-232 '57. (KIRE 10:?)

1. Institut geologii rudnykh mestorozhdeniy, petrografii, mineralogii i geokhimii AN SSSR, Moskva.
(Sphene)

SAVITSKIY, Ye.M.; TYLKINA, M.A.; PEKAREV, A.I.; CAVRILYUK, M.I.; ZABAVNOVA,
A.P.

Recrystallization diagram for cast tungsten. Dokl. AN SSSR 140
no.6:1301-1303 O '61. (MIRA 14:11)

1. Institut metallurgii im. A.A.Baykova AN SSSR. Predstavлено
академиком I.V.Tananayevym.
(Tungsten crystals--Growth)

STEPANOVA, M.N.; ODINOKOVA, V.A.; ZABAVSKAYA, E.A.

Neuroblastomas of the vertebral fissure in children.
Khirurgia no.9:81-85 '61. (MIRA 15:5)

1. Iz 2-oy khirurgicheskoy kliniki (zav. - prof. Ya.G. Dubrov),
patomorfologicheskogo (i. o. zav. A.A. Naumova) i rentgeno-
logicheskogo (zav. - dotsent A.I. Petrov) otdelov Moskovskogo
oblastnogo nauchno-issledovatel'skogo klinicheskogo instituta
imeni M.F. Vladimirovskogo.

(NERVOUS SYSTEM—TUMORS)

BARSUKOV, L.N., kand. sellekhnosyayatvennykh nauk; ZARAVSKAYA, E.M., nauchnyy
sotrudnik; IVANOVA, T.I., nauchnyy sotrudnik

Importance of turning over furrows. Zemledelie 7 no.11/67-71
N '59 (MIRA 13:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut udobreniy i
agropochvovedeniya.
(Plowing)

ZABAVSKAYA, K. M.

USSR/Biology - Ultrasonics

Sep/Oct '83

"Effects of High-Frequency Oscillations on Germination of Seeds and Development of Plants," L. N. Barsukov, Cand Agr Sci, and K. M. Zabavskaya

Agrobiol, No 5 (83), pp 80-85

Brief exposure (1-3 minutes) of seeds of cultivated plants to powerful mech oscillations of sonic frequency, has the same effect as exposure to ultrasonic oscillations, i.e. it accelerates germination, facilitates more rapid plant development and hastens

27612

maturity. High-frequency oscillations are of particular value in the cultivation of crops whose seeds are characterized by retarded and imperfect germination and development. Illustrated by charts.

RASKATOV, Afanasiy Ivanovich, dots.; ZARAVSKIY, A.V., nauchnyy red.;
CHISLOV, M.M., red.; PERSON, M.N., tekhn. red.

[Laboratory work in electrical engineering] Laboratoriye raboty
po elektrotekhnike. Moskva, Proftekhizdat, 1962. 326 p.
(MIRA 15:7)

1. Kafedra elektrotekhniki i elektroniki Moskovskogo tekhnologicheskogo instituta myasnoy i molechnoy promyshlennosti (for Raskatov).
(Electric engineering—Handbooks, manuals, etc.)
(Electric laboratories—Handbooks, manuals, etc.)

ZABAVSKIY, M.P., mayor, voyennyy letchik pervogo klassa

By authority of the wing commander. Vest.Vozd.Fl. no.8:23-
25 Ag '60. (MIRA 13:9)
(Russia--Air force) (Military discipline)

HEDNARZ, Stanislaw; ZABAWA, Mieczyslaw

Problem of traveling crane skewing. Hutnik P 29 no. 7/8; 264-269
JL-Ag '62.

1. Katedra Mechaniki Technicznej, Akademia Gorniczo-Hutnicza,
Krakow.

ZABAWA, Mieczyslaw, mgr inz., adiunkt

Way of measuring the average value of friction resistance in
bearings. Przegl mech 23 no.12:331-333 25 Je '64.

1. Department of Machine Parts, Academy of Mining and Metallurgy,
Krakow.

SOV/137-58-11-22143

Translation from: Referativnyy zhurnal. Metallurgiya, 1958, Nr 11, p 45 (USSR)

AUTHORS: Sorokin, P. Ya., Zabaykin, A. V., Babich, P. P., Zakharov, O. A.

TITLE: Continuous Measurement of the Temperature of Liquid Steel in the Ladle (Nepreryvnnyy zamer temperatury zhidkoy stali v kovshe)

PERIODICAL: Prom-ekon. byul. Sov. nar. kh-va Sverdl. ekon. adm. r-na, 1958, Nr 4, pp 3-6

ABSTRACT: The measurements are made in ladles of 30-45 t capacity by Pt/Ph-Pt thermocouple introduced into the ladle either by a dummy stopper from above or through the nozzle of the spare pouring aperture in the bottom of the ladle. The thermocouple junction is protected by covers made on a Zr-oxide base and are installed at 200-300 mm from the ladle bottom. The experiments conducted showed the temperature of the metal (Me) in the ladle, when under an adequate layer of slag, drops not at a gradient of 2-3°C/min, as had previously been held, but considerably more slowly. The method of continuous measurement of the temperature of the liquid steel makes it possible to determine the length of time during which the Me should be held in the ladle after the heat has been tapped, and this facilitates purification from nonmetallic

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SOV/137-58-11-22143

Continuous Measurement of the Temperature of Liquid Steel in the Ladle

and gas inclusions.

V. G.

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8(4)

SOV/32-24-12-21/45

AUTHORS: Sorokin, P. Ya., Zabaykin, A. V., Babich, F. P., Zakharov, G. A.

TITLE: Continuous Measurement of the Temperature of Molten Steel in
the Ladle (Neprezyvnyy zamer temperatury zhidkoy stali v
kovshe)

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, № 12, pp. 1475-1477
(USSR)

ABSTRACT: Immersion thermocouples give better results than optical
apparatus in the measurement of the temperature of molten
metals. From 1952 to 1954 continuous temperature measurements
were carried out in liquid steel still in ladles holding 30-
45 tons by the institute mentioned in the Association in col-
laboration with Ural'skiy vagonostroitel'nyy zavod (Ural
Car-Building Plant) and Zavod transportnogo mashinostroyeniya
v Chelyabinske (Transport Machine-Building Plant in Chelyabinsk).
The thickness of the lining of the ladles used was 200 mm
(walls) and 350 mm (floor). In one case the thermocouple was
mounted as a pseudo-seal (Fig 1), while in another case it
was introduced through the outlet (Fig 2). The experimental
results obtained (Figs 3-5) indicate the following: the

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Continuous Measurement of the Temperature of Molten Steel in the Ladle

temperature of the liquid metal becomes stable at a particular level after 15 minutes (curve of the figure). During the casting process the temperature of the liquid metal increases slowly in the case where a slag layer of 200-250 mm thick is present, or remains constant in the case where the slag layer is thinner. Contrary to wide-spread opinion, the temperature of the metal increases at the end of the casting process, and this finding agrees with the work of Van Gruyvenigen and Lauter (Ref 2), Pronov (Ref 3), Gruzin (Ref 4), and Boos and Vil'yams (Ref 5). The temperatures determined using optical pyrometers are always lower than those obtained using thermoelements. The temperatures in the upper metal layers are greater than in the lower layers (Figs 3,4). There are 5 figures and 5 Soviet references.

ASSOCIATION: Institut metallurgii Ural'skogo filiala Akademii nauk SSSR
(Institute of Metallurgy of the Ural Branch, Academy of Sciences, USSR)

Card 2/2

LORENTSO, D.N.; OKUNEV, I.V., inzh., red.; ZABAYKIN, A.Ya., inzh., red.;
KOZLOV, A.G., nauchnyy red.; MAREST'EV, M.I., red.; SUDOVICH, A.V., red.; YAMOV, A.F., red.; DUGINA, N.A., tekhn. red.

[Ural Railroad-Car Plant] Ural'skiy Vagonostroitel'nyy Zavod.
Moskva, Gos. nauchno-tekhn. izd-vo mashinostroit. lit-ry.
1961. 162 p. (MIRA 15:2)

1. Ural'skiy vagonostroitel'nyy zavod (for Lorentso).
(Nizhniy Tagil.—Railroads—Cars—Construction)

YUGOV, Vladimir Alekseyevich, kand.fiz.-mat. nauk; TELEGIN, R.V.,
doktor fiz.-mat. nauk, prof., reis.; ZABAZLAEVA, E.I.,
reis. [Redacted]

[thin films and their use in radio measuring techniques]
Tonkie plenki i ikh priznenie v radioizmernitel'noi tekhnike.
Moskva, Izd-vo Standartov, 1964. 122 p.
(MIRA 17:11)

"APPROVED FOR RELEASE: 03/15/2001

CIA-RDP86-00513R001963310018-9

OLEYNIK, Boris Nikolayevich; ZABOZLAYEVA, E.I., red.

[Exact calorimetry] Technika kalorimetrii. Moscow,
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